Zdravkovich, V.


Prince George's Community Coll., Largo, Md.

41p.; For related Packages 1-17, see SE 021 572-588; Not available in hard copy due to copyright restrictions

Prince George's Community College Bookstore, Largo, Maryland 20870 ($17.00 a set, $1.00 each)

MF-$0.83 Plus Postage. HC Not Available from EDRS.

*Autoinstructional Aids; *Chemistry; *College Science; Higher Education; *Independent Study; Individualized Instruction; Individualized Programs; *Organic Chemistry; Science Education; Self Help Programs

This booklet, one of a series of 17 developed at Prince George's Community College, Largo, Maryland, provides an individualized, self-paced undergraduate organic chemistry instruction module designed to augment any course in organic chemistry but particularly those taught using the text "Organic Chemistry" by Morrison and Boyd. The entire series of modules covers the first 13 chapters of the Morrison-Boyd text in great detail. Each module has been provided with from one to three audiotapes, available from Prince George's Community College, to provide students additional explanations of particular concepts. Each module includes a self-evaluation exercise, a reference guide, worksheets to be completed with the audiotapes, answer sheets for the worksheets, a progress evaluation, an answer sheet for the progress evaluation, an answer sheet for the self-evaluation exercise, an introduction to the topic covered by the module, and student performance objectives for the module. The topic of this module is alkanes. (SI)
ALKANES-NOMENCLATURE

1. I am butane
2. I am ethane
3. I am pentane
4. I am isohexane
5. I am heptane
Self Instructional Sequence in

ORGANIC CHEMISTRY

"Copr.," V. Zdravkovic 1976
Good morning, troublemaker -
I'm told they call you "Alkane".
Quite an impressive moniker,
Such a strange sounding name.

Actually, this is only your title,
Pretensions to grandeur, no less!
With aliases you've not been idle,
How do I keep track of this mess?!

Butane, pentane, hexane, heptane...
Won't you ever stop putting us on?
Your list of henchmen is legion
You're a species they can't seem to restrain
Methyl, ethyl, propyl, butyl...
As if you ground them out of a mill!

They say I'm to examine the textbook,
And listen to this marvelous tape,
And they've even come up with a system
Which holds out a small ray of hope.
The IUPAC nomenclature - designed to put you in your place.
I'll beat you at your own game you stinker -
I'll call you all-those-impossible-names right to your face!

---Author unknown
Definitions

The student will be able to define or explain and illustrate with appropriate examples where applicable the following terms: PARENT CHAIN, BRANCH, ALKYL GROUP, METHYL GROUP, ETHYL GROUP, \textit{n-PROPYL} GROUP, ISOPROPYL GROUP, \textit{n-BUTYL} GROUP, SEC. BUTYL GROUP, ISOBUTYL GROUP, TERT, BUTYL GROUP.

Naming

The student will be able to assign the correct IUPAC name to any given alkane.

The student will be able to draw the structure that corresponds to the given IUPAC name.

Given an incorrect name, the student will be able to draw the structure and assign the \textit{correct} IUPAC name to the compound.
Self Instructional Package No. 4
Form B - Self-Evaluation Exercise

ALKANES

NOMENCLATURE

Fill out this exercise by blacking the appropriate answers.

1. The longest chain in the compound below is:

```
C
/|
C-C-C C-C-C
/ |
C-C-C-C-C-C
/ |
C-C-C C-C-C
/|
C-C-C
```

a) 5  b) 6  c) 7  d) 8

2. The compound below is a derivative of:

```
C
/|
C-C-C
/ |
C-C-C-C-C-C
/ |
C-C-C C-C-C
/|
C
```

a) hexane  b) heptane  c) octane  d) nonane

3. You are given four different alkyl groups below. Identify the isobutyl group.

a) 
```
CH₃
\___/
CH₃-C-CH₃
```

c) 
```
CH₃
\___/
CH₃-CH₂-CH-
```

b) 
```
CH₃
\___/
CH₃-CH-CH₂-
```

d) 
```
CH₃-CH₂-CH₂-CH₂-
```

6  5
4. From the given alkyl groups select the secondary butyl group.

a) CH₃ CH(CH₃) CH₂
b) CH₃ CH₂ CH(CH₃)
c) CH₃

d) CH₃ CH₂ CH₂ CH₂

5. The correct IUPAC name for the compound with the structural formula drawn below is:

```
CH₃
|   CH₂-CH₃
|   CH₃CH₂-CH-CH₃
|   CH₃-C-CH₂-CH₂-C-CH₂-CH₃
|   CH₃-CH-CH₃
```

a) 5,5-dimethyl-2-ethyl-2-isopropyl-6-isobutyl heptane

b) 2,4,5,5-tetramethyl-8-ethyl-8-isopropyl nonane
c) 2,3,6,6,7,9-hexamethyl-3-ethyl decane
d) 2,4,5,5,8,9-hexamethyl-8-ethyl decane

6. The correct IUPAC name for the compound with the structural formula drawn below is:

```
CH₃ CH₃—CH₂—CH₂—CH₃
|   |   CH₃—CH—CH₂—CH₂—CH—CH₂—CH₃
|   |   CH₃ CH—CH₃
|   |   CH₂
|   |   CH₃
```

a) 3-methyl-7-n-propyl-6-sec. butyl-4-tert. butyl nonane

b) 2,2-dimethyl-6-n-propyl-3,5-disec. butyl octane
c) 3-methyl-7-ethyl-4-tert. butyl-6-sec. butyl decane
d) 2,2-dimethyl-6-ethyl-3,5-disec. butyl nonane
7. Identify the correct structural formula which corresponds to

2,4,4,6-tetramethyl-5-ethyl-5-isopropyl octane

a) \[
\begin{align*}
CH_3 & \quad CH \quad CH_2 \quad C \quad C \quad CH \quad CH_2 \quad CH_3 \\
& \quad CH_3 \quad CH_2\quad CH_2 \quad CH_3
\end{align*}
\]

b) \[
\begin{align*}
CH_3 & \quad CH \quad CH_2 \quad C \quad C \quad CH_2 \quad CH_2 \quad CH_3 \\
& \quad CH_3 \quad CH - CH_3
\end{align*}
\]

c) \[
\begin{align*}
CH_3 & \quad CH_2 \quad CH_2 \quad C \quad C \quad CH_2 \quad CH \quad CH_3 \\
& \quad CH_3 \quad CH \quad CH_3 \\
& \quad CH_3
\end{align*}
\]

d) \[
\begin{align*}
CH_3 & \quad CH_2 \quad CH_2 \quad C \quad C \quad CH_2 \quad CH \quad CH_3 \\
& \quad CH_2 \quad CH_3 \\
& \quad CH_2 - CH_3
\end{align*}
\]
8. Identify the structural formula which corresponds to:

3-ethyl-6-isopropyl-4-tert. butyl-5-isobutyl nonane

\[ \text{CH}_3 \]
\[ \text{CH}_3 \]
\[ \text{CH}_3 \]
\[ \text{CH}_3 \]
\[ \text{CH}_3 \]
\[ \text{CH}_3 \]

9
9. The correct IUPAC name for the alkane below is:

\[ \text{CH}_3\text{CH(\text{CH}_3)} \text{CH[CH(\text{CH}_3)}_2 \text{CH[CH}_2\text{CH(\text{CH}_3)}_2 \text{CH}_2 \text{C(\text{CH}_3)}_2 \text{CH}_3 \]

a) 2,6,6-trimethyl-3-isobutyl-4-tert. butyl heptane
b) 2,2,6-trimethyl-5-isopropyl-4-tert. butyl heptane
c) 2,2,6-trimethyl-5-n-propyl-4-tert. butyl heptane
d) 2,2,6-trimethyl-5-isopropyl-4-isobutyl heptane

10. From the given INCORRECT IUPAC name for an alkane draw its structural formula. The correct name of this alkane is:

ethyl, propyl, isopropyl, sec. butyl methane

a) 2-methyl-3-ethyl-3-sec. butyl hexane
b) 3-methyl-4-ethyl-4-isopropyl heptane
c) 2,4-dimethyl-3-ethyl-3-n-propyl hexane
d) 3-methyl-4-ethyl-4-n-propyl hexane
The reference guide should be used in conjunction with Form B or the Self Evaluation Exercise. The references given are geared specifically toward the questions on Form B.

Question 1
Question 2
Questions 3, 4
Questions 5, 6, 7, 8, 9, 10

Morrison and Boyd *Organic Chemistry*

For Questions 1 through 10, additional explanation and examples can be found in Tape 1 - Alkanes - Nomenclature.
OBJECTIVE: To learn the IUPAC nomenclature of ALKANES.

IUPAC RULES FOR THE NAMING OF ALKANES.
1. Select as the parent structure the longest chain, and then consider the compound to have been derived from this structure by the replacement of hydrogen by various alkyl groups.

2. Where necessary, indicate by a number the carbon to which the alkyl group is attached.

3. In numbering the parent carbon chain, start at whichever end results in the use of the lowest numbers.

4. If the same alkyl group occurs more than once as a side chain, indicate this by the prefix di-, tri-, tetra-, etc., to show how many of these alkyl groups there are, and indicate by various numbers the positions of each group.

5. If there are several different alkyl groups attached to the parent chain, name them in order of increasing size or in alphabetical order.

Example 1.

\[ \text{Longest chain: } 3 \text{- propane} \]
\[ \text{Number of branches: } 1 \]
\[ \text{Identity of the branch: } \text{methyl group} \]

Name: METHYL PROPANE

Example 2.

\[ \text{Longest chain: } 6 \text{- hexane} \]
\[ \text{Number of branches: } \]
\[ \text{Identity of the branch: } \]
Example 3.

\[
\begin{array}{c}
-\text{C- C-} \\
\text{C - C - C - C - C - C - C -} \\
1 2 3 4 5 6 7 \text{ CORRECT} \\
7 6 5 4 3 2 1 \text{ INCORRECT}
\end{array}
\]

**ASSIGNMENT 1.** a) Identify the longest carbon chain. b) Identify the branches. c) Assign the correct IUPAC names to the alkanes below.

I
\[
\begin{array}{c}
-\text{C- C- C-} \\
\text{C - C - C - C - C - C -} \\
\end{array}
\]

a) 

b) 

c) 

II
\[
\begin{array}{c}
-\text{C- C-} \\
\text{C - C -} \\
\text{C - C - C - C - C -} \\
\end{array}
\]

a) 

b) 

c) 

13
ASSIGNMENT 1. continued

Example 4.

ASSIGNMENT 2. Assign the correct IUPAC names to the following compounds:
ASSIGNMENT 2, continued

ASSIGNMENT 3. Draw the structural formulas of the following compounds:

I 2,2,3,4 - Tetramethyl Heptane

II 2,3,3,5,6 - Pentamethyl Nonane

III 2,2,3 - Trimethyl Butane

Example 5.

Longest chain: ____________
Number of branches: _________
Identity of branches: __________
IUPAC name: _______________

15
CH₄ Methane  - C – or CH₃  Methyl group
C₂H₆ Ethane  - C – C – or C₂H₅  Ethyl group
CₙH₂n + 2 Alkane  CₙH₂n + 1  Alkyl group

**Example 6.** 4 - ? propyl Octane

\[
\begin{align*}
&\text{I } \quad \text{C - C - C ( n-propyl group)} \\
&\quad \quad \text{I } \quad \text{C - C - C - C - C - C - C} \\
&\text{II } \quad \text{4-n-propyl Octane} \\
&\quad \quad \text{C - C - C ( n-propyl group)} \\
&\quad \quad \text{II } \quad \text{C - C - C - C - C - C - C} \\
&\text{III } \quad \text{4-n-propyl Octane} \\
&\quad \quad \text{C - C - C ( iso-propyl group)} \\
&\quad \quad \text{III } \quad \text{C - C - C - C - C - C - C} \\
&\text{4-iso-propyl Octane}
\end{align*}
\]

**ASSIGNMENT 4.** Write the structural formulas or the following compounds:

I 2,2- Dimethyl - 3 - ethyl - 4 - iso-propyl Heptane

II 2,4,5- Trimethyl - 3,3 - diethyl Octane

III 2,3,7,8- Tetramethyl - 3 - isopropyl - 4-n-propyl Nonane
ASSIGNMENT 4. continued

IV 4,4- Di-isopropyl Octane

ASSIGNMENT 5. Name the compounds for which the structural formulas are given below:

\[ \text{I} \quad \text{II} \quad \text{III} \]

\[
\begin{align*}
\text{I} & \quad \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} \\
\text{II} & \quad \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} \\
\text{III} & \quad \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C}
\end{align*}
\]
Forgetful Frieda has been given the task of assigning the correct IUPAC names to the alkanes below. In each case she has forgotten something. Examine the structural formulas and correct Frieda's incorrectly assigned IUPAC names.

2,6,7- Pentamethyl - 5 - ethyl - 4 - propyl Octane

2,2,4- Trimethyl - 5,6 - di - n-propyl Heptane

2 - Methyl - 4 - isopropyl - 3 - ethyl Heptane
Confused Clyde has completely failed the assignment given him by his instructor. He was given a structural formula and asked to supply the correct IUPAC name. Confused Clyde not only returned an incorrect answer, but he also managed to lose the original structural formula. It is now your task to draw the structural formula of the compound from the incorrect name which Clyde submitted and to supply the correct IUPAC name for it.

Name submitted by Confused Clyde was:

Methyl, ethyl, isopropyl Methane

Example 7.

\[ \begin{array}{c}
\text{n-Butane} & \text{C - C - C - C} \\
\text{1} & \text{2} & \text{1} & \text{n-Butyl group} \\
\end{array} \]

\[ \text{5 - n - Butyl Nonane:} \]

\[ \begin{array}{c}
\text{C - C - C - C} \\
\text{sec. Butyl group} \\
\end{array} \]

\[ \text{5 - sec. Butyl Nonane:} \]
Example 7. continued

Isobutane $\text{C} - \text{C} - \text{C} - \text{C} - \text{C}$

Isobutyl group

tert. Butyl group

5 - Isobutyl Nonane

5 - tert. Butyl Nonane

ALKYL GROUPS $\text{C}_n\text{H}_{2n+1}$

Methyl group $\text{CH}_3$

Ethyl group $\text{CH}_3 - \text{CH}_2$

n-Propyl group $\text{CH}_3 - \text{CH}_2 - \text{CH}_2 - \text{CH}_3$

Isopropyl group $\text{CH}_3 - \text{CH} - \text{CH}_3$ or $\text{CH}_3 - \text{C}$

n-Butyl group $\text{CH}_3 - \text{CH}_2 - \text{CH}_2 - \text{CH}_2$

sec. Butyl group $\text{CH}_3 - \text{CH}_2 - \text{CH} - \text{CH}_3$ or $\text{CH}_3 - \text{CH}_2 - \text{CH}$

$\text{CH}_3$

20

19
Assign the correct IUPAC names to the four compounds listed below.

I \[\text{C-C-C-C-C-C} \]

II \[\text{C-C-C-C-C-C-C} \]

III \[\text{C-C-C-C-C-C} \]
Inert Irma has been given the task of assigning the IUPAC names to a number of alkanes. She has done this only partially correctly. She has actually made a few mistakes. You are now expected to assume the role of a teacher and mark her answers as right or wrong. In addition to this, whenever an answer is incorrect, show the correction.

2,2 - Dimethyl - 4 - isopropyl - 5 - tert. butyl - 7 - iso-butyl Octane

2,5 - Dimethyl - 3 - ethyl - 2 - n-propyl - 4 - sec. butyl Hexane
SELF INSTRUCTIONAL PACKAGE 1
Tape 1 - Worksheet cont.

ASSIGNMENT 9. continued

ASSIGNMENT 10. From the given IUPAC names write the structural formulas of
the alkanes below.

I 2,2,6 - Trimethyl - 5 - isopropyl - 3 - tert. butyl Octane

II 2,4 - Dimethyl - 5 - ethyl - 3 - isopropyl - 4 - tert. butyl Heptane

III 2,3,6 - Trimethyl - 5 - isopropyl - 4 - sec. butyl Nonane
ASSIGNMENT 11.

Saturated Sam has completed an assignment given him, and has submitted IUPAC names for three alkanes. These names, however, are incorrect. Using these incorrect IUPAC names:

a) draw the correct structural formulas
b) assign the correct IUPAC names.

I  Isobutyl, isopropyl, ethyl, methane

II  2 - n - propyl - 3 - sec. butyl - 4 - isobutyl Pentane

III  2 - tert. butyl - 3 - isopropyl - 4 - sec. butyl Hexane
Self Instructional Package 4
Tape 1 - Answer Sheet

ALKANES

Assignment 1. a) Identify the longest carbon chain. b) Identify the branches. c) Assign the correct IUPAC names to the alkanes below:

\[
\begin{align*}
&\text{I} \quad \text{Octane} \\
&\text{II} \quad \text{Three methyl groups} \\
&\text{III} \quad \text{2,3,5-Trimethyl Octane}
\end{align*}
\]

Assignment 2. Assign the correct IUPAC names to the following compounds:

\[
\begin{align*}
&\text{I} \quad \text{Hexane} \\
&\text{II} \quad \text{Two methyl groups} \\
&\text{III} \quad \text{2,3-Dimethyl Hexane}
\end{align*}
\]

\[
\begin{align*}
&\text{I} \quad \text{Nonane} \\
&\text{II} \quad \text{Three methyl groups} \\
&\text{III} \quad \text{2,4,6-Trimethyl Nonane}
\end{align*}
\]

\[
\begin{align*}
&\text{I} \quad 2,2,5\text{-Trimethyl Heptane}
\end{align*}
\]
ASSIGNMENT 2. continued

2,4,6,6 - Tetramethyl Octane

2,2,5,8 - Tetramethyl Nonane

ASSIGNMENT 3. Draw the structural formulas of the following compounds:

I 2,2,3,4 - Tetramethyl Heptane

\[
\begin{align*}
\text{CH}_3 & \quad \text{CH}_3 & \quad \text{CH}_3 \\
1 & \quad 3 & \quad 3 & \quad 3 \\
\text{CH}_3 & \quad \text{C} & \quad \text{CH} & \quad \text{CH} & \quad \text{CH} & \quad \text{CH}_2 & \quad \text{CH}_3 \\
& \quad 1 & \quad \text{CH}_3
\end{align*}
\]

II 2,3,3,5,6 - Pentamethyl Nonane

\[
\begin{align*}
\text{CH}_3 & \quad \text{CH}_3 & \quad \text{CH}_3 & \quad \text{CH}_3 \\
1 & \quad 3 & \quad 1 & \quad 3 \\
\text{CH}_3 & \quad \text{C} & \quad \text{CH} & \quad \text{CH} & \quad \text{CH} & \quad \text{CH}_2 & \quad \text{CH}_3 & \quad \text{CH}_3 \\
& \quad 1 & \quad \text{CH}_3
\end{align*}
\]

III 2,2,3 - Tetramethyl Butane

\[
\begin{align*}
\text{CH}_3 & \quad \text{CH} \\
1 & \quad 3 & \quad 3 \\
\text{CH} & \quad \text{C} & \quad \text{CH} & \quad \text{CH} \\
& \quad 1 & \quad \text{CH}_3 & \quad \text{CH}_3
\end{align*}
\]
ASSIGNMENT 4. Write the structural formulas of the following compounds:

I 2,2 - Dimethyl - 3 - ethyl - 4 - isopropyl Heptane

\[ \text{C} - \text{C} - \text{C} \]
\[ \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} \]
\[ \text{C} - \text{C} \]

II 2,4,5 - Trimethyl - 3,3 - diethyl Octane

\[ \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} \]
\[ \text{C} - \text{C} \]

III 2,3,7,8 - Tetramethyl - 3 - isopropyl - 4 - n-propyl Nonane

\[ \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} \]
\[ \text{C} - \text{C} \]

IV 4,4 - Di-isopropyl Octane

\[ \text{C} - \text{C} - \text{C} \]
\[ \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} \]

ASSIGNMENT 5. Name the compound for which the structural formulas are given below:

\[ \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} \]
\[ \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} \]

3,5-Dimethyl - 4 - ethyl - 6 - isopropyl Octane
ASSIGNMENT 5. continued

2,2,6,7- Tetramethyl - 6- ethyl - 4- isopropyl Octane

2,3,6,6- Tetramethyl - 5- ethyl - 5- n-propyl Octane

Forgetful Frieda has been given the task of assigning the correct IUPAC names to the alkanes below. In each case she has forgotten something. Examine the structural formulas and correct Frieda's incorrectly assigned IUPAC names.

2,2,6,6,7- Pentamethyl - 5- ethyl - 4- isopropyl Octane
CONFUSED CLYDE HAS COMPLETELY FAILED THE ASSIGNMENT GIVEN HIM BY HIS INSTRUCTOR. HE WAS GIVEN A STRUCTURAL FORMULA AND ASKED TO SUPPLY THE CORRECT IUPAC NAME. CONFUSED CLYDE NOT ONLY RETURNED AN INCORRECT ANSWER, BUT HE ALSO MANAGED TO LOOSE THE ORIGINAL STRUCTURAL FORMULA. IT IS NOW YOUR TASK TO DRAW THE STRUCTURAL FORMULA OF THE COMPOUND FROM THE INCORRECT NAME WHICH CLYDE SUBMITTED, AND TO SUPPLY THE CORRECT IUPAC NAME FOR IT.

NAME SUBMITTED BY CONFUSED CLYDE WAS:

METHYL, ETHYL, ISOPROPYL METHANE
ASSIGNMENT 7. continued

\[
\begin{align*}
4 & \quad \text{CH}_2 - \text{CH}_3 \\
\text{STRUCTURE:} & \quad \begin{array}{c}
\text{CH} \quad 3 \\
\text{C} \quad 2 \\
\text{CH}_2 \quad 1 \\
\text{H} \quad \text{CH}_3
\end{array}
\end{align*}
\]

CORRECT IUPAC NAME: 2,3- Dimethyl Pentane

ASSIGNMENT 8. Assign the correct IUPAC names to the four compounds listed below:

\[
\begin{align*}
\text{I} & \quad \text{C-C-C-C-C-C-C} \\
\text{II} & \quad \text{C-C-C-C-C-C-C-C} \\
\end{align*}
\]

2,2,5- Trimethyl - 3 - isopropyl - 4 - tert. butyl Heptane

2,3,7- Trimethyl - 4 - ethyl - 5 - iso- butyl - 6 - sec, butyl Nonane
ASSIGNMENT 8, continued

2.3.7.7- Tetramethyl - 4 isopropyl - 5 - isobutyl Nonane

2.3.7- Trimethyl - 4 - ethyl - 5 - tert, butyl Octane

ASSIGNMENT 9.

Inert Irma has been given the task of assigning the IUPAC names to a number of alkanes. She has done this only partially correctly. She has actually made a few mistakes. You are now expected to assume the role of a teacher and mark her answers as right and wrong. In addition to this, whenever an answer is incorrect, show the correction.
ASSIGNMENT 9, continued

2,2,7,9-Tetramethyl-4-isopropyl-5-tert. butyl Decane

3,6,6-Trimethyl-5-ethyl-4-isopropyl Nonane

2,2,6-Trimethyl-4-ethyl-5-isopropyl-3-tert. butyl Octane
ASSIGNMENT 10. From the given IUPAC names, write the structural formulas of the alkanes below.

I

\[
\begin{align*}
\text{C} & \quad \text{C} \quad \text{C-C-C} \\
\text{C-C-C} & \quad \text{C} \\
\text{C-C} & \quad \text{C} \\
\end{align*}
\]

II

\[
\begin{align*}
\text{C} & \quad \text{C-C-C} \\
\text{C-C-C} & \quad \text{C} \\
\text{C-C} & \quad \text{C-C} \\
\text{C} & \quad \text{C} \\
\end{align*}
\]

III

\[
\begin{align*}
\text{C} & \quad \text{C-C-C} \\
\text{C-C-C} & \quad \text{C} \\
\text{C} & \quad \text{C} \\
\text{C} & \quad \text{C} \\
\text{C} & \quad \text{C} \\
\end{align*}
\]
Saturated Sam has completed an assignment given him and submitted IUPAC names for three alkanes. These names, however, are incorrect. Using these incorrect IUPAC names:

a) draw the correct structural formulas
b) assign the correct IUPAC names.

I Isobutyl, isopropyl, ethyl, methane (INCORRECT)

CORRECT IUPAC NAME: 2,5-Dimethyl - 3-ethyl Hexane

II 2-n-propyl - 3-sec. butyl - 4-isobutyl Pentane (INCORRECT)

CORRECT IUPAC NAME: 2,4,6-Trimethyl - 5-sec. butyl Nonane

III 2-tert. butyl - 3-isopropyl - 4-sec. butyl Hexane (INCORRECT)

CORRECT IUPAC NAME: 2,2,3,6-Tetramethyl - 5-ethyl - 4-isopropyl Octane
ALKANES

NOMENCLATURE

1. The parent chain in the compound below consists of how many carbon atoms?

![Compound Diagram]

a) 7  b) 8  c) 9  d) 10

2. The compound below can be considered a derivative of which alkane?

![Compound Diagram]

a) butane  b) hexane  c) heptane  d) octane

3. Identify the isobutyl group as one of the groups listed.

a) \((\text{CH}_3)_3\text{C}\)

b) \text{CH}_3 \text{CH(\text{CH}_3)} \text{CH}_2

c) \text{CH}_3 \text{CH}_2 \text{CH(\text{CH}_3)}

d) \text{CH}_3 \text{CH}_2 \text{CH}_2 \text{CH}_2
4. Assign the correct IUPAC name to the alkane for which the structural formula is supplied.

\[
\begin{align*}
\text{CH}_3 & \quad \text{CH}_3 \\
\text{CH}_3 & \quad \text{CH}_2 \quad \text{CH} \\
\text{CH}_3 & \quad \text{CH}_2 \quad \text{CH}_2 \quad \text{C} \quad \text{CH}_2 \quad \text{CH}_3 \\
\text{CH}_3 & \quad \text{CH} \quad \text{CH}_3 \\
\text{CH}_3 & \quad \text{CH}_3 \\
\end{align*}
\]

a) 3-methyl-5-isopropyl-2-tert. butyl-5-isobutyl heptane
b) 2,2,3,4-tetramethyl-6-isopropyl-6-isobutyl octane
c) 2,2,3,4,8-pentamethyl-6-ethyl-6-isopropyl nonane
d) 2,2,3,4,8-pentamethyl-6-ethyl-6-propyl nonane

5. Assign the correct IUPAC name to the alkane for which the structural formula is given below.

\[
\begin{align*}
\text{CH}_3 & \quad \{ \\
\text{CH}_3 & \quad \text{CH}_2 \quad \text{CH} \\
\text{CH}_3 & \quad \text{CH}_2 \quad \text{C} \quad \text{CH}_2 \quad \text{CH}_3 \\
\text{CH}_3 & \quad \text{CH} \quad \text{CH}_3 \\
\text{CH}_3 & \quad \text{CH}_3 \\
\end{align*}
\]

a) 2,3,3,5,7-pentamethyl-5-ethyl octane
b) 2,4,6-trimethyl-4-ethyl-2-isopropyl heptane
c) 2,4-dimethyl-2-isopropyl-4-isobutyl hexane
d) 2,3,3,5,7-pentamethyl-5-ethyl nonane
6. The correct IUPAC name which corresponds to the alkane below is:

\[ \text{CH}_3 \text{CH(CH}_3) \text{CH(CH}_3) \text{[C(CH}_3)_3] \text{CH(} \text{CH}_2 \text{CH}_3 \text{)} \text{CH(} \text{CH}_2 \text{CH}_3 \text{)} \text{CH}_2 \text{CH}_3 \]

a) 3,4-diethyl-6-isopropyl-5-tert. butyl heptane
b) 2,3-dimethyl-5,6-diethyl-4-tert. butyl octane
c) 6,7-dimethyl-2,3-diethyl-4-tert. butyl octane
d) 2,3-dimethyl-5,6-diethyl octane

7. The correct IUPAC name corresponding to the alkane below is:

\[ \text{CH}_3 \text{CH[CH}_2 \text{CH(CH}_3)_2] \text{CH(CH}_3) \text{CH[CH(CH}_3)_2]CH(} \text{CH}_2 \text{CH}_3 \text{)} \text{CH}_2 \text{CH}_2 \text{CH}_3 \]

a) 2,4,5-trimethyl-6-isopropyl nonane
b) 2,2,4,5-tetramethyl-6-isopropyl octane
c) 2,4-dimethyl-5-isopropyl octane
d) 2,4,5-trimethyl-5-propyl nonane

8. The structural formula which corresponds to 2,3-dimethyl-6-ethyl-5-isopropyl-4-tert. butyl octane is:

a) \[ \text{CH}_3 \text{CH(CH}_3) \text{CH(CH}_3) \text{CH[CH(CH}_3)_3] \text{CH(} \text{CH}_2 \text{CH}_2 \text{CH}_3 \text{)} \text{CH}_2 \text{CH}_2 \text{CH}_3 \]

b) \[ \text{CH}_3 \text{CH(CH}_3) \text{CH(CH}_3) \text{CH[CH(CH}_3)_3] \text{CH(} \text{CH}_2 \text{CH}_2 \text{CH}_3 \text{)} \text{CH}_2 \text{CH}_2 \text{CH}_3 \]

c) \[ \text{CH}_3 \text{CH(CH}_3) \text{CH(CH}_3) \text{CH[CH(CH}_3)_3] \text{CH(} \text{CH}_2 \text{CH}_2 \text{CH}_3 \text{)} \text{CH}_2 \text{CH}_2 \text{CH}_3 \]

d) \[ \text{CH}_3 \text{CH(CH}_3) \text{CH(CH}_3) \text{CH[CH(CH}_3)_3] \text{CH(} \text{CH}_2 \text{CH}_2 \text{CH}_3 \text{)} \text{CH}_2 \text{CH}_2 \text{CH}_3 \]

9. From the incorrect IUPAC name for an alkane which is:

1-ethyl-1-isopropyl-2-isobutyl ethane, draw the structure and assign the correct IUPAC name. It is:

a) 2-methyl-5-isopropyl-heptane
b) 2,6-dimethyl heptane
c) 2,6-dimethyl-3-ethyl heptane
d) 2,6-dimethyl-5-ethyl heptane
10. From the **incorrect** IUPAC name for an alkane which is:

    2-isobutyl-2-tert. butyl-3-n-butyl-3-sec. butyl butane

   Draw the structure and assign the correct IUPAC name. It is:

a) 2,4,5-trimethyl-3-tert. butyl-5-sec. butyl nonane
b) 2,2,3-trimethyl-3-isobutyl-4-sec. butyl octane
c) 2,2,3,5-tetramethyl-3-isobutyl-4-n-butyl heptane
d) 2,4,5,6-tetramethyl-4-tert. butyl-5-n-butyl nonane
ALKANES

NOMENCLATURE

1. d
2. d
3. b
4. b
5. c, d
6. c
7. c
8. d
9. d
10. b
ALKANES

NOMENCLATURE

1. d
2. d
3. b
4. c
5. a
6. b
7. a
8. b
9. c
10. a
S.I.P. #4 - ERRATA

FORM D - Question 10

a) 2,4,5-Trimethyl-4-tert. Butyl-5-Sec. Butyl Nonane

TAPE 1 - Answer Sheet

Assignment 8

Compound II - 2,3,3,7 - Tetramethyl-4-Ethyl-5-Isobutyl-6-Sec. Butyl Nonane

Compound IV - 2,3,3,7 - Tetramethyl-4-Ethyl-5-tert. Butyl Octane.